

## **GALVANIC REACTION CHART**

Below is a galvanic reaction chart for dissimilar metals.

This chart is designed to assist in broadly assessing the risk of galvanic corrosion associated with a given metal coming into contact with another metal. To use the chart, align the metal to be assessed (for the risk of corrosion) in the left column with the Contact Metal listed in the upper row; green represents a lower risk\* and red represents a higher risk. For a more specific assessment of the risk of galvanic corrosion, please reference other sources.

			Contact Metal													
Galvanic Corrosion Risk		Magnesium and Alloys	Zinc and Alloys	Aluminum and Alloys	Cadmium	Carbon Steels	Cast Iron	Stainless Steel	Lead, Tin, and Alloys	Nickel	Brasses, Nickel-Silvers	Copper	Bronzes, Cupro-Nickels	Nickel Copper Alloys	Nickel-Chrome Alloys, Titanium, Silver, Graphite, Gold, and Platinum	
Corroding Metal	Magnesium and Alloys															
	Zinc and Alloys															
	Aluminum and Alloys															
	Cadmium															
	Carbon Steel															
	Cast Iron															
	Stainless Steels															
	Lead, Tin, and Alloys															
	Nickel															
	Brasses, Nickel-Silvers															
	Copper															
	Bronzes, Cupro-Nickels															
	Nickel Copper Alloys															
	Nickel-Chrome Alloys, Titanium, Silver, Graphite, Gold, and Platinum															

\*Green represents "lower risk" not "no risk."

It should be noted that if sacrificial plating is incorporated into the fastener design, then galvanic action can result in the deterioration of the sacrificial coating, rather than of the fastener. We would advise that the suggested fasteners for dissimilarmetal applications would incorporate our GrabberGard<sup>®</sup> coating that utilizes both barrier and sacrificial coatings to minimize the chance and/or rate of corrosion. The barrier coating used to encapsulate the base steel and zinc layer plus anti-corrosion chemical bonding agents minimize the opportunity for contact to occur, thereby further minimizing the risk of corrosion.